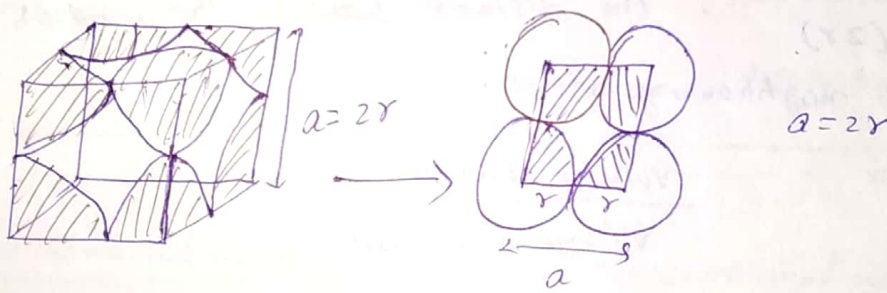


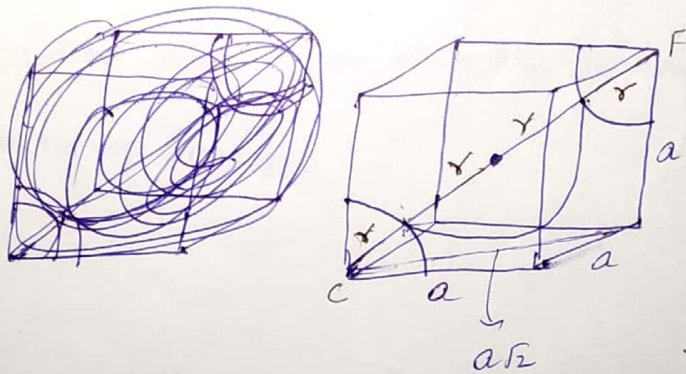
Characteristics of cubic lattice:-

	Simple c	BCC	FCC
Volume, Conventional cell	a^3	a^3	a^3
Lattice point per cell	1	2	4
Volume, Primitive cell	a^3	$a^3/2$	$a^3/4$
Lattice point per unit vol.	$1/a^3$	$2/a^3$	$4/a^3$
Number of nearest neighbours	6	8	12
Nearest neighbour distance	a	$\frac{\sqrt{3}a}{2}$	$\frac{a}{\sqrt{2}}$
Number of Second neighbours	12	6	6
Second neighbour distance	$\sqrt{2}a$	a	a
Packing fraction	$\frac{\pi}{6} = 0.524$	$\frac{\pi\sqrt{3}}{8} = 0.680$	$\frac{\pi\sqrt{2}}{6} = 0.740$

① Simple cubic



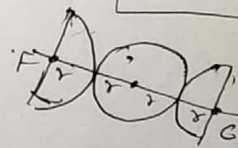
② BCC



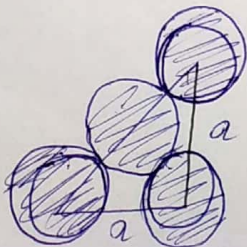
$$(4r)^2 = a^2 + (a\sqrt{2})^2$$

$$(4r)^2 = 3a^2$$

$$r = \frac{a\sqrt{3}}{4}$$



③ FCC



$$2a^2 = (4r)^2 \Rightarrow r = \frac{\sqrt{2}a}{4}$$

Hexagonal close Packed (hcp) structure :-

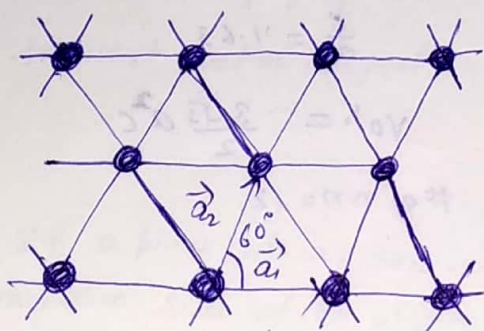
In fcc & hcp packing factor = 0.74

hcp example \rightarrow Be, Cd, Os ...

ABAB -- Stacking hcp

ABCABC -- Stacking fcc

First Simple Hexagonal Bravais lattice \rightarrow

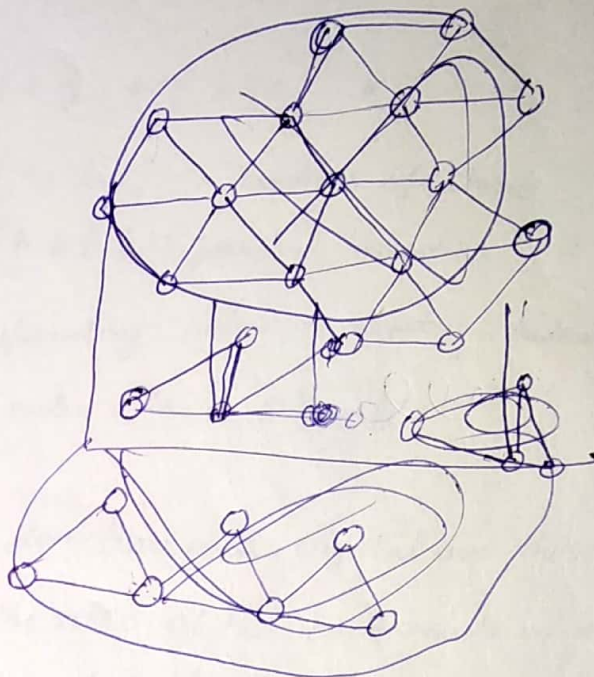


$$|\vec{a}_1| = |\vec{a}_2| = a$$

$$\begin{cases} \vec{a}_1 = a\hat{x}, \vec{a}_2 = \frac{a}{2}\hat{x} + \frac{\sqrt{3}a}{2}\hat{y} \\ \vec{a}_3 = c\hat{z} \end{cases}$$

defines hex

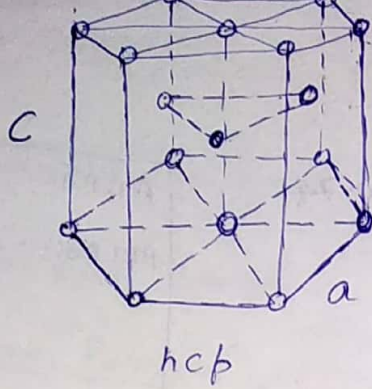
first two primitive vectors generate a triangular lattice in the x-y plane, and the third stacks the planes a distance c above one another.



hcp \rightarrow two interpenetrating simple hexagonal Bravais lattice

displaced from one another by $\frac{\vec{a}_1}{3} + \frac{\vec{a}_2}{3} + \frac{\vec{a}_3}{c}$

Consider stacking cannonballs, starting with a close-packed triangular lattice as the first layer. The next layer is formed by placing a ball in the depression left in the centre of every other triangle in the first layer, forming second triangular layer shifted wrt first. Third layer formed by placing balls in alternate depressions in the second layer, so that they lie directly over the balls in the first layer. Fourth layer directly on the second layer and so on.



$$n = \frac{3}{2} + \frac{3}{2} + 3 = 6$$

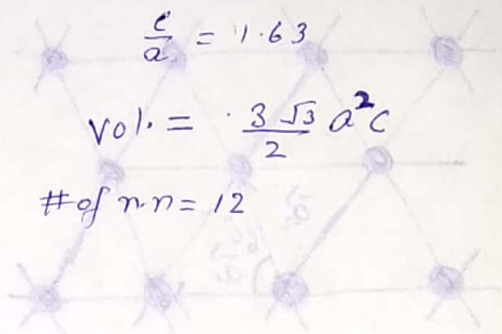
atoms touch each other along the edge of the hexagon

$$\Rightarrow a = 2r$$

$$\frac{c}{a} = 1.63$$

$$\text{Vol.} = \frac{3\sqrt{3}}{2} a^2 c$$

$$\# \text{ of } n \cdot n = 12$$



$$a = \sqrt{3}r = 1.732r$$

$$\vec{a}_1 = a \hat{x}, \vec{a}_2 = \frac{a}{2} \hat{x} + \frac{\sqrt{3}a}{2} \hat{y}, \vec{a}_3 = \sqrt{3} \hat{y}$$

$$\vec{a}_3 = \sqrt{3} \hat{y}$$

Substitution

first two primitive vectors are in the

the x-y plane, and the third vector is

above or another

hcp → two interpenetrating simple hexagonal lattices

displaced from each other by $\frac{1}{2} \vec{a}_1 + \frac{1}{2} \vec{a}_2$

Consider stacking (concentric) spheres in a close-packed

structure. Let us take the first layer as the reference. The next layer is formed

by placing a ball in the depression left in the center of any

three spheres in the first layer, forming a second layer.

Next, a third layer is formed by placing balls

in the depressions in the second layer. The third layer

is directly over the balls in the first layer.

Miller Indices

(i) Intercepts of plane on a_1, a_2, a_3 axes are $3a_1, 2a_2, 2a_3$

(ii) $\frac{3a_1}{a_1}, \frac{2a_2}{a_2}, \frac{2a_3}{a_3}$

$\equiv 3, 2, 2$

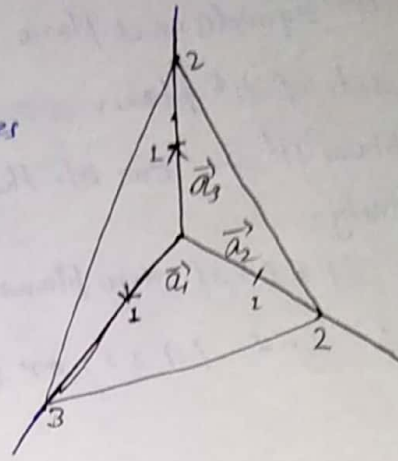
(iii) Reciprocal $\rightarrow \frac{1}{3}, \frac{1}{2}, \frac{1}{2}$

(iv) Smallest set of integral number $6 \times \frac{1}{3}, 6 \times \frac{1}{2}, 6 \times \frac{1}{2} =$

$= (2, 3, 3) \rightarrow$ Indices of plane

$(hkl) \rightarrow$ general notation

(v) If a plane cuts an axis on the negative side of the origin, the corresponding index is negative, indicated by placing a minus sign above the index: (h, \bar{k}, l) etc.



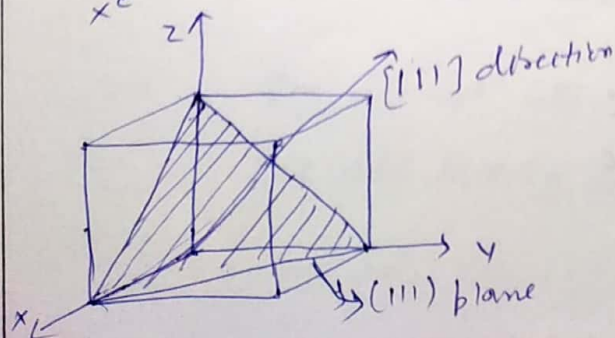
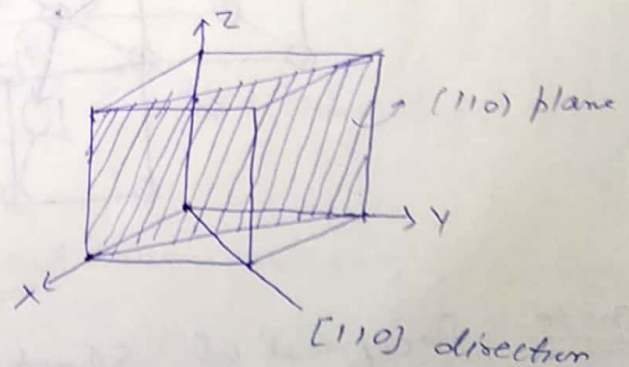
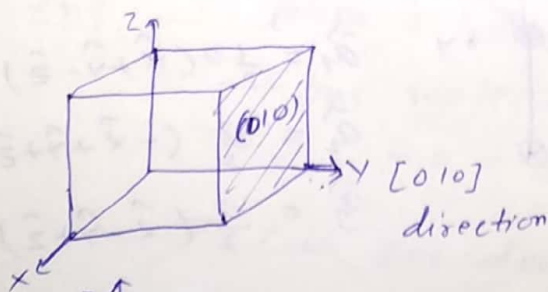
Direction

Indices $[uvw]$ of a direction in a crystal are the set of the smallest integers that have the ratio of the components of a vector in the desired direction, referred to the axes.

\vec{a}_1 axis is $[100]$ direction

$-\vec{a}_2$ axis is $[0\bar{1}0]$ direction.

In Cubic crystal direction $[hkl]$ is \perp^{an} to a plane (hkl) but this is not generally true in other crystals.



Note (i) All hkl equidistant planes have same $(hkl) \Rightarrow (hkl)$ defines a set of hkl planes.

(ii) A plane hkl to one of the coordinate axes has an intercept at infinity.

(iii) If (hkl) of two planes have the same ratio, i.e., $(8,4,4)$ and $(4,2,2)$ or $(2,1,1)$, then planes are hkl to each other.

Separation between lattice planes in Cubic crystal -

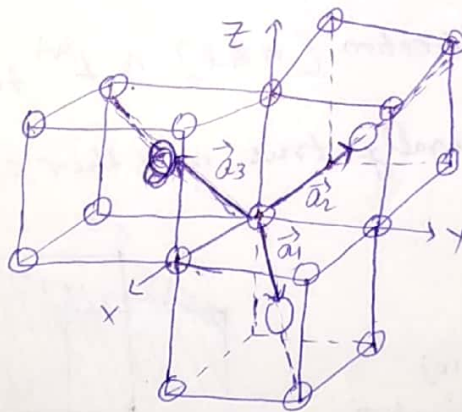
$$d = \frac{a}{\sqrt{h^2 + k^2 + l^2}}$$

Lattice vectors in Cubic system

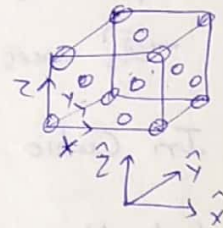
SC $\rightarrow \vec{a}_1 = a\hat{x}, \vec{a}_2 = a\hat{y}, \vec{a}_3 = a\hat{z}$

fcc $\rightarrow \vec{a}_1 = \frac{a}{2}(\hat{y} + \hat{z}), \vec{a}_2 = \frac{a}{2}(\hat{z} + \hat{x}), \vec{a}_3 = \frac{a}{2}(\hat{x} + \hat{y})$

bcc \rightarrow



$$\begin{aligned} \vec{a}_1 &= \frac{1}{2}a(\hat{x} + \hat{y} - \hat{z}) \\ \vec{a}_2 &= \frac{a}{2}(-\hat{x} + \hat{y} + \hat{z}) \\ \vec{a}_3 &= \frac{a}{2}(\hat{x} - \hat{y} + \hat{z}) \end{aligned}$$



note Reciprocal of
 SC \rightarrow SC
 bcc \rightarrow fcc
 fcc \rightarrow bcc